# Elements

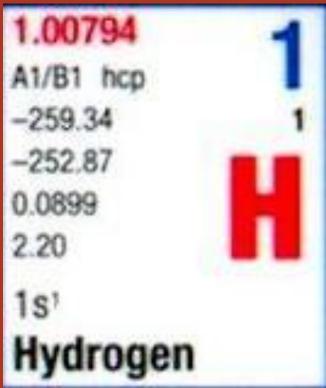


# Chemical symbols

Usually consist of one or two letters

Ex: H -> Hydrogen Hg ->

Mercury



#### **Chemical Properties**

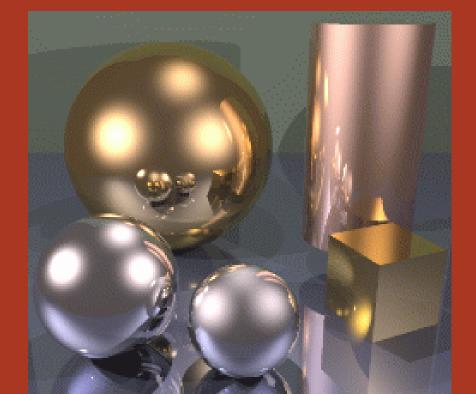
 Describes a substance's ability to react chemically with other substances to form new products



#### Metals

Solid, shiny, malleable and ductile,

Good conductors of heat and electricity



#### **Non-metals**

 Usually gaseous or fragile solids, especially when at room temperature



Chlorine gas

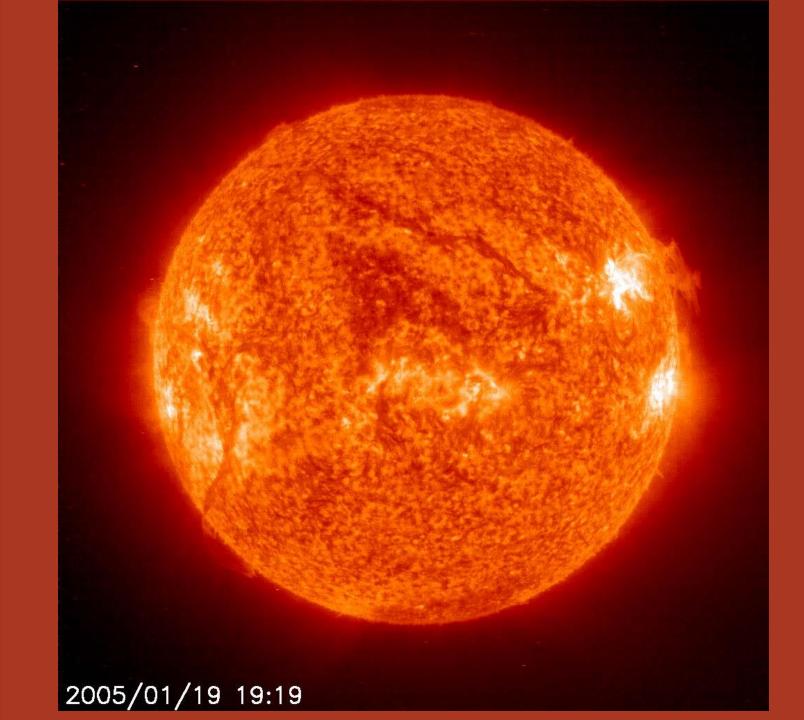
#### Hydrogen (H)

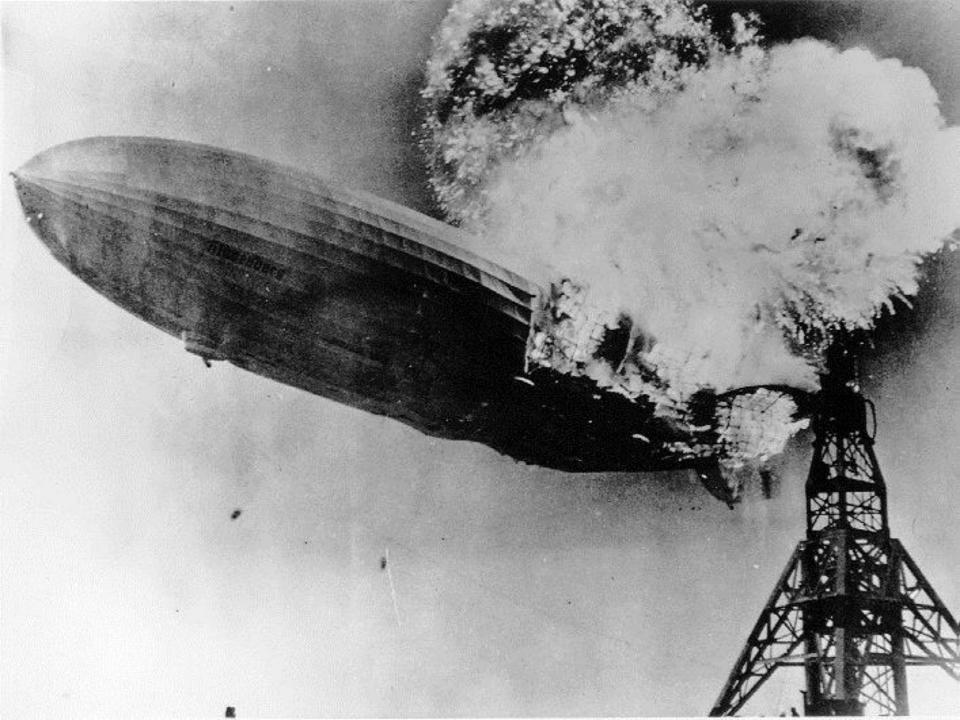
- Colourless, odourless, tasteless
- Highly flammable and reactive
- Makes up most of the Sun as welll as other stars
- Makes up over 90% of the atoms in the universe
- Most of the H on Earth combines with O to form H<sub>2</sub>0
- Less dense than air → used to inflate weather balloons

# Hydrogen (H)

lonized hydrogen







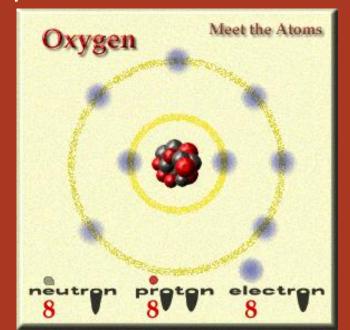
#### Iron (Fe)

- Robust metal (even more so when combined with carbon → steel
- Rusts when combined with O<sub>2</sub> and H<sub>2</sub>0 (oxidization)



#### Oxygen (O)

- Non-metal
- Necessary for all life
  - Used to burn up sugar to provide energy in the cell
- Makes up 21% of our atmosphere
- For the past 3 billion years, oxygen has been produced by plants



### Sodium (Na)

- Unique metal because of its softness
- Reacts violently in water and produces a toxic gas.
  - » Sodium + water gives off a large quantity of energy and sodium gas in the form of an explosion





#### Chlorine (CI)

- Yellowish-green gas
- Toxic in large quantities
- Used as a disinfectant (pools, hot tubs)





# What happens when you combine sodium and chlorine? Keep in mind that these are both toxic!

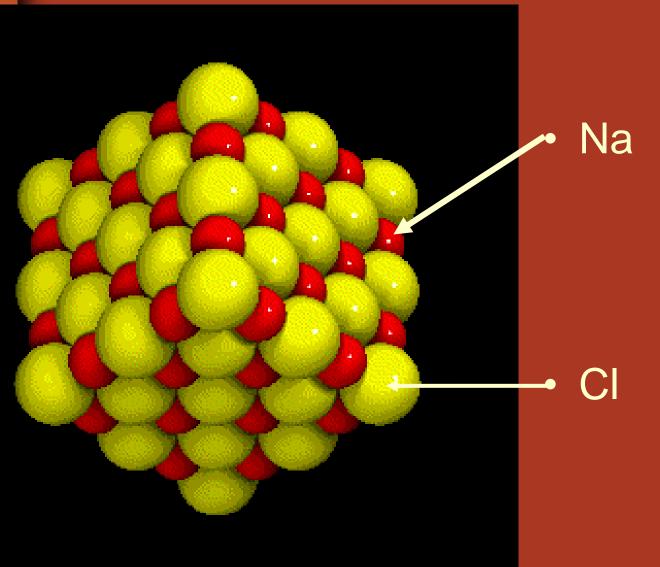




#### Why does salt take up a cubic shape?



## NaCl (table salt)



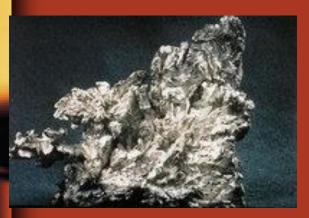
### Mercury (Hg)

- Toxic metal
- The only metal that is liquid at room temperature



#### Silver (Ag)

- White metal with several physical properties that make it useful and commercial
- Ex: jewellery, money, musical instruments, technological instruments







#### Silicon

- 2<sup>nd</sup> most abundant element in the Earth's crust (Oxygen is the most common)
- A métalloïd
- A semi-conducteur (conducts electricity, but poorly conducts heat)
- Used to build computer chips and components
- With oxygen, forms quartz and opal



